

ERRORS OF MEASUREMENT OF THE CALORIFIC VALUE OF COMBUSTIBLE GASES IN A BOMB CALORIMETER

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Abstract: The volumetric (specific) heat (energy) of combustion (volumetric calorific value — VCV) of natural combustible gases is an important, practically demanded value. An analysis was made of the sources of errors in the measurements of VCV according to the Russian state standard methods using a calorimeter for burning with a bomb. The contribution of the accuracy of the instruments used to measure the VCV is quantified. It is shown that the existing methods for measuring VCV do not take into account the calorimetric features of gaseous fuel combustion. A method is proposed for measuring the VCV of natural combustible gases that takes these features into account. Data are presented on the measurements of the VCV of the reference methane with an estimate of the error in the results obtained, confirming the effectiveness of the proposed method.

Keywords: higher volumetric heat of combustion; lower volumetric heat of combustion; natural gas; bomb calorimeter for burning, measurement procedure; temperature; pressure; measurement error

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