

HOMOGENEOUS PYROLYSIS OF DIMETHYL ETHER UNDER PULSED ADIABATIC COMPRESSION

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Abstract: Thermal decomposition of dimethyl ether (DME) has been studied in a rapid compression machine over a temperature range 1060–1920 K. The main products (H_2 , CO, CH_4 , and formaldehyde) and minor products of reaction have been identified. Some of them like propane, propene, allene, methylacetylene, butadiene-1,3, vinylacetylene, diacetylene, cyclopentadiene, benzene and toluene were identified for the first time. It is shown that ethylene yield remains constant in the temperature range 1700–1920 K at the conversion degree of DME exceeding 95% along with the decrease of methane and ethane yields and the increase of acetylene yield.

Keywords: dimethyl ether (DME); pyrolysis; rapid compression machine (RCM); formaldehyde; ethylene

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References

- Gayvoronskiy, A. I., V. A. Markov, and Yu. V. Ilatovskiy. 2007. *Ispol'zovanie prirodnogo gaza i drugikh al'ternativnykh topliv v dizel'nykh dvigatelyakh* [Utilization of natural gas and other alternative fuels in Diesel engines]. Moscow: Gazprom Publ. 480 p.
- Park, S. H., and C. S. Lee. 2013. Combustion performance and emission reduction characteristics of automotive DME engine system. *Prog. Energ. Combust.* 39(1):147–168. doi: 10.1016/j.pecc.2012.10.002.
- Khadzhiev, S. N., A. L. Maksimov, and M. V. Krotova. 2017. Proizvodstvo dimetilovogo efira: tsel' i vozmozhnosti organizatsii v Rossii s uchetom geopoliticheskogo faktora [The production of dimethyl ether: The purpose and scope of the organization in Russia, given the geopolitical factor]. *Nauchn. zh. Ross. gazovogo obshchestva* [Scientific J. of the Russian Gas Society] 1:43–53.
- Vólnina, E. A., M. A. Kipnis, and S. N. Khadzhiev. 2017. Catalytic chemistry of dimethyl ether (review). *Pet. Chem.* 57(5):353–373. doi: 10.1134/S0965544117050139.
- Zhao, Z., M. Chaos, A. Kazakov, and F. L. Dryer. 2008. Thermal decomposition reaction and a comprehensive kinetic model of dimethyl ether. *Int. J. Chem. Kinet.* 40(1):1–18. doi: 10.1002/kin.20285.
- Sivaramkrishnan, R., J. V. Michael, A. F. Wagner, et al. 2011. Roaming radicals in the thermal decomposition of dimethyl ether: Experiment and theory. *Combust. Flame* 158(4):618–632. doi: 10.1016/j.combustflame.2010.12.017.
- Leifer, E., and H. C. Urey. 1942. Kinetics of gaseous reactions by means of the mass spectrometer. The thermal decomposition of dimethyl ether and acetaldehyde. *J. Am. Chem. Soc.* 64(4):994–1001. doi: 10.1021/ja01256a070.
- Lossing, F. P., K. U. Ingold, and A. W. Tickner. 1953. Free radicals by mass spectrometry. Part II. — The thermal decomposition of ethylene oxide, propylene oxide, dimethyl ether, and dioxane. *Dissc. Faraday Soc.* 14:34–44. doi: 10.1039/DF9531400034.
- Anderson, K. H., and S. W. Benson. 1962. Termination products and processes in the pyrolysis of dimethyl ether. *J. Chem. Phys.* 36(9):2320–2323. doi: 10.1063/1.1732883.
- Aronowitz, D., and D. Naegeli. 1977. High-temperature pyrolysis of dimethyl ether. *Int. J. Chem. Kinet.* 9(3):471–479. doi: 10.1002/kin.550090314.
- Hidaka, Y., K. Sato, and M. Yamane. 2000. High-temperature pyrolysis of dimethyl ether in shock waves. *Combust. Flame* 123(1):1–22. doi: 10.1016/S0010-2180(00)00122-X.
- Pyun, S. H., W. Ren, K.-Yu. Lam, et al. 2013. Shock tube measurements of methane, ethylene and carbon monoxide time-histories in DME pyrolysis. *Combust. Flame* 160(4):747–754. doi: 10.1016/j.combustflame.2012.12.004.
- Benson, S. W. 1956. Pyrolysis of dimethyl ether. *J. Chem. Phys.* 25(1):27–31. doi: 10.1063/1.1742841.
- Kolbanovskiy, Yu. A., V. S. Shchipachev, N. Ya. Chernyak, et al. 1982. *Impul'snoe szhatie gazov v khimii i tekhnologii* [Impulsive compression of gases in chemistry and technology]. Moscow: Nauka. 240 p.
- Bilera, I. V., and N. N. Buravtsev. 2016. Gomogennyy piroliz izopentana v usloviyakh adiabaticeskogo szhatiya [The homogeneous pyrolysis of isopentane under pulsed adiabatic compression]. *Goren. Vzryv (Mosk.) — Combustion and Explosion* 9(1):74–82.

16. Bilera, I. V. 2017. Vysokotemperaturnyy gomogennyy piroliz etana v reaktore adiabaticheskogo szhatya [The high-temperature homogeneous pyrolysis of ethane in the adiabatic compression reactor]. *Goren. Vzryv (Mosk.) — Combustion and Explosion* 10(2):12–17.

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