

# EFFECT OF STEAM ON THE DYNAMICS OF NO FORMATION DURING CH<sub>4</sub> COMBUSTION IN THE PERFECTLY STIRRED AND PLUG FLOW REACTORS

G. A. Poskrebyshev, A. N. Ermakov, I. A. Korobeinikova, and V. N. Popov

V. L. Talrose Institute of Energy Problems of Chemical Physics, Russian Academy of Sciences, 38-2 Leninsky Prosp., Moscow 119334, Russian Federation

**Abstract:** This work is devoted to the calculation of the dynamics of NO formation during combustion of the premixed air–methane stoichiometric mixtures ( $\phi = 1$ ) in the perfectly stirred and the plug flow (PFR) reactors at the presence of steam. It is shown that in both types of the reactors, the increase of the H<sub>2</sub>O/CH<sub>4</sub> ratio from 0 to 5 reduces the concentration of the emitted NO. The power of the relative decrease of NO concentration followed the increasing of H<sub>2</sub>O/CH<sub>4</sub> ratio does not depend on the reactor type. However, it is found that the absolute concentrations of NO, determined at the time corresponding to the maximum temperature, are always lower in the case of PFR and is due to the lower residence time of the combustion products in the high-temperature zone. According to this study, the kinetics of NO formation in PFR can be described by a two-staged process called in the present work as the “fast” and “slow” reactions that correspond, respectively, to the flame and postflame zones. It is also shown that the effect of steam, added to the initial mixture, on the rate of chemical reactions producing NO is significant only in the case of PFR and only at the initial time (fast) of combustion ( $\leq 30$  ms). At the same time, it is shown that the concentration of molecular nitrogen in the initial mixture also affects the chemistry of NO formation, but in both stages.

**Keywords:** methane; steam; combustion; NO; perfectly stirred reactor; plug flow reactor

## Acknowledgments

The work was supported by the Division of Energetics, Mechanical Engineering, Mechanics, and Operation Control and Management of the Processes of the Russian Academy of Sciences under program “Thermophysical problems in construction and operation of the high-efficiency gas power machines of new generation.”

## References

1. Parmon, V. N., Z. R. Ismagilov, O. N. Favorskij, A. A. Belokon, and V. M. Zaharov. 2007. Primenenie kataliticheskikh kamer sgoraniya v gazoturbinnnykh ustanovkakh detsentralizovannogo energosnabzheniya [Application of catalytic combustion chambers in gas turbine plant for decentralized energy supply]. *Her. Russ. Acad. Sci.* 77(9):819–826.
2. Polezhaev, Ju. V., R. R. Grigor’janc, and A. N. Ermakov. 2009. Parogazovye ustanovki monarnogo tipa. Problemy i perspektivy sozdaniya [Steam–gas installations of monar type. Problems and prospects of creation]. *Energetika Tatarstana* [Power Engineering of Tatarstan] 1:6–14.
3. Smith, G. P., D. M. Golden, M. Frenklach, N. W. Moriarty, B. Eiteneer, M. Goldenberg, C. T. Bowman, R. K. Hanson, S. Song, W. C. Gardiner, Jr., V. V. Lissianski, and Z. Qin. 1999. GRI-Mech 3.0. Available at: [combustion.berkeley.edu/gri-mech/version30/text30.html](http://combustion.berkeley.edu/gri-mech/version30/text30.html) (accessed October 9, 2017).
4. Zhao, D., H. Yamashita, K. Kitagawa, N. Arai, and T. Furuhashi. 2002. Behavior and effect on NO<sub>x</sub> formation of OH radical in methane–air diffusion flame with steam addition. *Combust. Flame* 130:352–360.
5. Kobayashi, H., Y. Soichiro, Y. Iechikawa, and Y. Ogami. 2009. Dilution effects of superheated water vapor on turbulent premixed flames at high pressure and high temperature. *Proc. Combust. Inst.* 32:2607–2614.
6. Wang, F., J. Mi, and P. Li. 2013. Combustion regimes of a jet diffusion flame in hot co-flow. *Energ. Fuel.* 27:3488–3498.
7. Titova, N. S., P. S. Kuleshov, O. N. Favorskii, and A. M. Srarik. 2014. The features of ignition and combustion of composite propane–hydrogen fuel: Modeling study. *Int. J. Hydrogen Energ.* 39:6764–6773.
8. Zel’dovich, Ya. B., P. Ya. Sadovnikov, and D. A. Frank-Kamenetskii. 1947. *Okislenie azota pri gorenii* [Oxidation of nitrogen during combustion]. Ed. N. N. Semenov. Moscow–Leningrad: Publishing House of the USSR Academy of Sciences. 148 p.

Received February 14, 2017

## Contributors

**Poskrebyshev Gregory A.** (b. 1965) — Candidate of Science in chemistry, leading scientist, V. L. Talrose Institute of Energy Problems of Chemical Physics, Russian Academy of Sciences, 38-2 Leninsky Prosp., Moscow 119334, Russian Federation; gposkr@chph.ras.ru

**Ermakov Alexander N.** (b. 1943) — Doctor of Science in chemistry, head of laboratory, V. L. Talrose Institute of Energy Problems of Chemical Physics, Russian Academy of Sciences, 38-2 Leninsky Prosp., Moscow 119334, Russian Federation; polclouds@yandex.ru

**Korobeinikova Irina A.** (b. 1962) — Candidate of Science in chemistry, senior scientist, V. L. Talrose Institute of Energy Problems of Chemical Physics, Russian Academy of Sciences, 38-2 Leninsky Prosp., Moscow 119334, Russian Federation; koriren@rambler.ru

**Popov Vladislav N.** (b. 1930) — senior scientist, V. L. Talrose Institute of Energy Problems of Chemical Physics, Russian Academy of Sciences, 38-2 Leninsky Prosp., Moscow 119334, Russian Federation; vnpop@rambler.ru