

# CERIUM FLUORIDES: INFLUENCE ON BORON OXIDATION AND NEW METHOD OF SYNTHESIS

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**Abstract:** The aim of this work is increasing the efficiency of oxidation of amorphous boron in air. The influence of cerium fluorides on suppression of the reduction reaction of boron has been studied. Thermodynamic analysis and experimental investigations (differential scanning calorimetry, thermogravimetric analysis, and thermometry by microthermocouples) of interaction between the cerium fluoride mixtures with boron and boron oxide have been carried out. The results of the research prove the possibility of decreasing the temperature of boron active oxidation in the presence of CeF<sub>3</sub> or CeF<sub>4</sub>, which is more preferable. The lower thermal stability of CeF<sub>4</sub> compared to CeF<sub>3</sub> has been found. An original method of synthesis of CeF<sub>4</sub> using XeF<sub>2</sub> fluorination has been proposed.

**Keywords:** cerium fluorides CeF<sub>n</sub>; cerium trifluoride CeF<sub>3</sub>; cerium tetrafluoride CeF<sub>4</sub>; amorphous boron; boron oxide B<sub>2</sub>O<sub>3</sub>; xenon difluoride XeF<sub>2</sub>; boron oxidation; thermal stability of cerium fluorides; thermodynamic (TD) calculations; differential scanning calorimetry (DSC); thermogravimetric analysis (TGA); exothermic effect; endothermic effect; CeF<sub>4</sub> synthesis; XRD analysis; analysis of experimental results

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Received February 09, 2017

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