

THE FORMATION ENTHALPIES AND RADICALS REORGANIZATION OF AZIDO-CONTAINING COMPOUNDS

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Abstract: Standard formation enthalpies of some azido-containing compounds in standard state and in gas phase are determined. The bond dissociation energies C–N are determined. Using fundamental equations of chemical physics, the calculation scheme of energies of reorganization of molecule fragments into radicals by a method of “double difference” is suggested. The new calculation method is proposed to determine the energies of reorganization of molecule fragment N₃ into radical N₃[•].

Keywords: enthalpies of formation and vaporization; “double difference” calculation method; radical; reorganization energy; bond energy

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